Abolishing activity against ascorbate in a cytosolic ascorbate peroxidase from switchgrass

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ABSTRACT

Switchgrass (Panicum virgatum L.) is being developed as a bioenergy species. Recently an early version of its genome has been released permitting a route to the cloning and analysis of key proteins. Ascorbate peroxidases (APx) are an important part of the antioxidant defense system of plant cells and present a well studied model to understand structure–function relationships. Analysis of the genome indicates that switchgrass encodes several cytosolic ascorbate peroxidases with apparent varying levels of tissue expression. A major cytosolic ascorbate peroxidase was thus selected for further studies. This gene was cloned and expressed in Escherichia coli to obtain purified active protein. Full heme incorporation of the enzyme was achieved utilizing slow growth and supplementing the media with 5-aminolevulinic acid. The enzyme was observed to be monomeric in solution via size exclusion chromatography. Activity toward ascorbate was observed that was non-Michaelis–Menten in nature. A site-directed mutant, R172S, was made in an attempt to differentiate activity against ascorbate versus other substrates. The R172S protein exhibited negligible ascorbate peroxidase activity, but showed near wild type activity toward other aromatic substrates.

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1. Introduction

Cytosolic ascorbate peroxidase (APx) catalyzes the reduction of H₂O₂ utilizing small organic substrates as electron donors with the physiological substrate being ascorbate (AsA) as shown below (Eq. (1)).

H₂O₂ + APX → 2H₂O + 2MDAsA

(1)

H₂O₂

In plants, the monodehydroascorbate radical (MDAsA) is converted back to AsA by MDASa reductase or in the absence of reductase, the 2 MDAsAs can disproportionate to AsA and dehydroascorbic acid (DAsA) (Chen et al., 2003; Shigeoka et al., 2002). The catalytic mechanism of ascorbate oxidation by APx has been well studied and is generally represented as shown in Eqs. (2)–(4), where APx first reacts with H₂O₂ and becomes Compound I, the fully oxidized form of APx. Compound I is converted back to APx via two one-electron transfers from an electron donating substrate (HS).

APX + H₂O₂ → Compound I + H₂O

(2)

Compound I + HS → Compound II + S⁺

(3)

Compound II + HS → APX + S⁺ + H₂O

(4)

There are many excellent reviews dealing with APx (Dunford, 1999; Raven, 2003; Raven et al., 2004), and only a brief review is provided here. Ascorbate-dependent peroxidases from spinach and pea were first described in 1979 (Groden and Beck, 1979; Kelly and Latzko, 1979) and are key proteins protecting cellular health in plants by scavenging peroxides generated during photosynthesis and cellular metabolism (Alvarez and Lamb, 1997; Bowler et al., 1992; Ishikawa and Shigeoka, 2008; Noctor and Foyer, 1998). They have been investigated at the structural and functional level because unlike cytochrome c peroxidase (CcP), one of the best known and characterized peroxidases, they have the more typical Compound I intermediate with a porphyrin-based radical and they utilize small molecule substrates as electron donors (Raven, 2003). Ascorbate-dependent peroxidases are essentially defined by having high specificity to ascorbate, which has a clear physiological role as an electron source for detoxifying cells of H₂O₂. However, many of these enzymes also have significant activity toward other aromatic electron donors, which usually do not have clear physiological roles. For the most part, these ascorbate-dependent peroxidases have higher activity toward ascorbate than other substrates.

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Crystal structures of APxs have been used to identify two different substrate binding sites, one for ascorbate (PDB ID: 1OAF) (Sharp et al., 2003) and one for aromatic substrates (PDB IDs: 1VOH, complexed with salicyhydroxamic acid and 2VCS, complexed with isoniazid) (Metcalfe et al., 2008; Sharp et al., 2004). This would strongly indicate that the substrate specificity for the other aromatic compounds comes from a different binding site than that of ascorbate.

In 1995, the first crystal structure of APx was determined for recombinant pea cytosolic APx (PDB ID: 1APX) (Patterson and Poulos, 1995). This structure showed a dimeric enzyme held together non-covalently by ionic interactions. The structure was remarkably similar to that of CcP, especially in the heme active site where the same hydrogen bonds exist between key conserved amino acids that are known to be essential for function. One striking difference was the discovery of a cation (K⁺) binding site that has been observed to influence the ligation of the distal histidine to the heme iron (Cheek et al., 1999). In 2003, another set of structures were determined for recombinant soybean cytosolic APx that included a structure of the APx-ascorbate complex (PDB ID: 1OAF) (Sharp et al., 2003). This structure positively identified the c-edge of the heme as the primary location of ascorbate binding and paved the way to better understanding the electron transfer mechanism for the enzyme. Presently, there are almost 30 APx structures in the protein structural database (http://www.rscb.org) that are predominantly mutant structures for cytosolic APx.
APx has proven to be a useful platform for protein engineering studies of peroxidases (Cheek et al., 1999). Most recently, APx has been engineered as a reporter for electron microscopy that can be used for imaging of mammalian organelles (Martell et al., 2012).

In the work reported here, a major cytosolic APx from switchgrass (rPviAPx) was cloned, expressed and its initial characterization performed; interestingly, switchgrass has been targeted as a biofuel crop (Sarath et al., 2008). Additionally, homology modeling was also performed using crystal structural models of APx and site-directed mutagenesis on a key amino acid found in the ascorbate binding site, Arg 172 (R172). This residue has been previously identified as essential for ascorbate binding such that even conservative size and charge changes like R172K dramatically lower activity toward ascorbate (Macdonald et al., 2006). This residue was thus mutated to a serine (R172S) both to validate our homology model and test the effect of placing a much smaller polar amino acid (previously untested mutation) in a position between the ascorbate and predicted aromatic substrate binding sites.

2. Results and discussion

2.1. Sequence analysis

Using the early release of the switchgrass genome (Draft0.0, http://www.phytozome.org), the fully sequenced PviAPx gene was found to be identical to a cytosolic ascorbate peroxidase encoded by the gene Pavirv00022559m.g. A BLAST (Altschul et al., 1990) analysis of this sequence showed that there were at least 12 other closely related sequences in the switchgrass genome (e-value threshold of $4.2 \times 10^{-31}$ or lower), leading us to speculate that there were a total of six APx genes each in the A and B genomes. APx can be nominally assigned to three different clades, comprising the chloroplastic, peroxisomal and cytosolic proteins (Teixeira et al., 2004). Similarly, available switchgrass APx protein sequences appeared to be clearly distinguishable into three clades (Fig. 1A). Comparison of the switchgrass sequences annotated as cytosolic APxs with six other cytosolic APx sequences obtained from the NCBI database was carried out, namely: Zea mays APx1 (NP_001152249.1); Sorghum bicolor APx1 (XP_002468053.1); Oryza sativa APx1 (NP_001049769.1), Puccinellia sativum APx01(AAA33645), Glycine max APx1 (NP_001237785.1), and Arabidopsis thaliana APx1 (NP_1772267.1). This was essential to validate the identification of the switchgrass cDNA clone as a cytosolic APx, as it is closely related to the other monocot cytosolic APx, and is also relatively closely related to the dicot sequences used for these analysis. To further confirm that this specific switchgrass APx is expressed in cells, a bioinformatic analysis was performed of relative transcript abundances in a number of different tissues (Fig. 1B). rPviAPx is abundantly expressed in most tissues obtained from several different switchgrass cultivars, suggesting that this enzyme is an important part of the switchgrass cellular antioxidant defense.

When compared to other recombinant APxs of solved crystal structures, it has a very high sequence similarity to both recombinant pea (rPsAPx01) and soybean (rGmAPx1) with 82% identity/90% positive and 85% identity/90% positive, respectively. This level of identity indicates that both of these enzymes would be useful for developing a homology structural model of rPviAPx.

2.2. rPviAPx characterization

Expression of holo rPviAPx depends on the simultaneous production of protein and heme by the bacteria. This requirement was met by supplying the bacteria with the heme precursor, 5-amino levulinic acid (5-ALA) and “slow growing” the cultures prior to induction with lower temperature (31°C) and gentler shaking (150 rpm). The cultures at this point generally had a reddish color when compared to cultures without the 5-ALA, which was interpreted as production of excess heme. Once protein expression was induced with isopropyl β-D-1-thiogalactopyranoside (IPTG), the newly expressed APx was able to fully incorporate heme due to this excess. Growing the protein this way allowed for a reproducible expression of holo rPviAPx for both WT and mutant. Protein purity was then characterized both by Rz (~2.25) and SDS-PAGE. In gels of the proteins, routine purification to a single band at ~26 kDa was achieved. Occasionally, it was observed that a faint impurity at about 54 kDa was present, which is possibly some kind of covalently linked dimer of the protein. Fig. 2 shows a typical SDS–PAGE of rPviAPx before and after thrombin cleavage. Lanes

Fig. 2. SDS–PAGE of Ni–NTA purified rPviAPx before and after thrombin cleavage of 6X-His tag. (1) Marker. (2) Desalted Ni–NTA purified before cleavage. (3) Thrombin cleavage reaction. (4) Purified protein collected by running cleavage reaction over Ni–NTA column. (5) 5× volume of lane 4 sample demonstrating relative purity.

Fig. 3. Spectra of reduced WT and mutant rPviAPx.

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2–4 show that the protein is successfully cleaved of its His tag, while lane 5 (same sample as lane 4 but with 5/C2 volume) shows that the purified protein is of real apparent homogeneity. Comparison of size exclusion chromatography (SEC) with SDS–PAGE results strongly indicates that the native state of rPviAPx is monomeric. The calculated size from the amino acid sequence for the His-tagged protein is 29.5 kDa (calculated using ProtParam from ExPASy) (Wilkins et al., 1999). The His-tagged protein has a size of approximately 26–27 kDa on the SDS–PAGE gel (Fig. 2). The chromatogram for the rPviAPx indicates one major species of 35 kDa with a minor species of 46 kDa. The major species is most likely the monomeric form, since it is relatively close to monomeric sizes determined from the denaturing gel (26 kDa) and the ProtParam calculation (29.5 kDa). The minor species of larger size is most likely an unfavored dimeric form of the protein. Pea and soybean cytosolic APx purified from plant sources are also reported to be homodimeric (Dalton et al., 1987; Mittler and Zilinskas, 1991). However, there is a report of a purified APx from Zea mays that was determined to be a 28 kDa momomeric protein via SEC (Koshiba, 1993). From the crystal structure of the recombinant pea APx (rpAPx), it was observed that there are seven residues involved in eight salt-bridge contacts. After aligning the amino acid sequence of rPviAPx to that of rpAPx, it was discovered that six of the seven residues responsible for dimerization are identical. The seventh residue is Asp in Pea but is Lys in rPviAPx. This single amino acid difference would result in a repulsive charge interaction is likely the cause of the loss of dimeric association for rPviAPx. Further inspection of the gene database allowed us to identify a gene for cytosolic APx for Zea mays (NP_001152249.1) that also has this same amino acid difference.

Spectra were collected for WT and mutant in resting, reduced and oxidized states. The reduced state spectra obtained from sodium dithionite reduced WT and mutant are shown in Fig. 3. In the Soret region (Fig. 3A), the WT protein had a resting peak maximum at 405 nm whereas the mutant was at 410 nm. When both proteins were reduced, the Soret peak shifted to 433 nm and 424 nm, respectively. In the 550 nm region of the spectra (Fig. 3B), significant differences between the WT and mutant enzyme spectra were observed. However, in an oxidized state generated by addition of H2O2, differences are less pronounced (Fig. 4). Of interest, in the oxidized state for the mutant there was a decrease in the intensity of the peak at 411 nm, and a red-shift was not observed, as compared to the WT enzyme (Fig. 4). Taken together, the spectra of the mutant suggest little spectral change in the transition from a resting to an oxidized state. In contrast, the spectra of the WT showed distinct shifts of 405–412 nm and 505–520 nm. These data suggest that the R172S mutant is shifted from the resting state toward the oxidized state prior to addition of H2O2 (Table 1).

<table>
<thead>
<tr>
<th>Enzyme – state</th>
<th>Wavelength maxima (nm)</th>
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<tbody>
<tr>
<td>WT – resting</td>
<td>405, 505, 539, 572, 637</td>
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<tr>
<td>WT – reduced</td>
<td>433, 509, 552, 582a</td>
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<td>412, 520, 550, 625</td>
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<td>R172S – resting</td>
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2.3. Enzyme activity

The activity of both WT and mutant enzyme was tested using ascorbate and various aromatic substrates. As expected, high activity was observed with ascorbate for the WT enzyme, but very little activity with the mutant. Since APx has been observed to display non-Michaelis–Menten kinetics (Lad et al., 2002), the kinetic constants, $K_e$ (becomes $K_n$ when $n = 1$) and $k_{cat}$ was determined by using the solver function in Excel to find the best-fit parameters for the Hill equation (Eq. (5)). Specifically, solver minimized the root mean square deviation between our data and values for WT and mutant in resting, reduced and oxidized states. The reduced state spectra obtained from sodium dithionite reduced WT and mutant are shown in Fig. 3. In the Soret region (Fig. 3A), the WT protein had a resting peak maximum at 405 nm whereas the mutant was at 410 nm. When both proteins were reduced, the Soret peak shifted to 433 nm and 424 nm, respectively. In the 550 nm region of the spectra (Fig. 3B), significant differences between the WT and mutant enzyme spectra were observed. However, in an oxidized state generated by addition of H2O2, differences are less pronounced (Fig. 4). Of interest, in the oxidized state for the mutant there was a decrease in the intensity of the peak at 411 nm, and a red-shift was not observed, as compared to the WT enzyme (Fig. 4). Taken together, the spectra of the mutant suggest little spectral change in terms of wavelength in transitioning from a resting to an oxidized state. In contrast, the spectra of the WT showed distinct shifts of 405–412 nm and 505–520 nm. These data suggest that the R172S mutant is shifted from the resting state toward the oxidized state prior to addition of H2O2 (Table 1).

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Fig. 4. Spectra of WT and mutant after addition of H2O2.
calculated using the Hill equation (which reduces to the Michaelis–Menten equation when $n = 1$) by changing $V_{\text{max}}$, $K$, and $n$. Non-Michaelis–Menten kinetics of WT rPviAPx toward ascorbate was observed (Fig. 5B). This corresponds well to what has been reported for recombinant soybean APx, where strong evidence of two ascorbate binding sites is observed and a non-hyperbolic relationship is observed between velocity and substrate concentration (Lad et al., 2002). The kinetic parameters used to find the best fit where $K^0 = 804 \pm 168 \mu M$ and $k_{\text{cat}} = 161 \pm 43 \mu M/s$, where $n = 1.6$ and error is 2 SD for a series of three microplate assays. Because of the very low activity of the mutant, it is difficult to measure the kinetic parameters but the data does fit well to a hyperbolic (Michaelis–Menten kinetics) function, although the error bars are quite large.

$$\frac{v}{V_{\text{max}}} = \frac{[S]^n}{K^0 + [S]^n}$$

(5)

Both the WT and mutant enzymes have activity toward the aromatic compound, 2,2’-azino-bis(3-ethylbenzothiazoline-6-sulfonic acid) (ABTS). Fig. 5A shows a plot of the data for both enzymes, where the same enzyme concentration is used but different $k_{\text{cat}}$ and $K^0$ values measured. $k_{\text{cat}}$ of $0.40 \pm 0.022 \text{s}^{-1}$ and $K^0$ of $43.4 \pm 4.04 \text{nM}$ ($n = 1$) for WT and a $k_{\text{cat}}$ of $0.67 \pm 0.093 \text{s}^{-1}$ and $K^0$ of $110 \pm 16 \text{nM}$ ($n = 1$) was observed for the mutant enzyme. However, this activity is much lower than the ascorbate activity. This is somewhat consistent with the relative level of ABTS activity versus ascorbate activity in other APxs, where it has been observed to be as low as 3% of the ascorbate activity (Dalton et al., 1996; Sajitha Rajan and Murugan, 2010). Although, the activity appears to be different between the WT and mutant, the difference is not very large and the mutation does not appear to have a significant effect on ABTS activity compared to WT.

Initial attempts to develop microplate assays for guaiacol and pyrogallol proved to be difficult. Progress curves for each substrate where the substrate concentration was held constant, but the concentration of enzyme was varied indicated that the enzyme is inactivated by the oxidized guaiacol and pyrogallol products. Progress curves for guaiacol, pyrogallol and ABTS (data not shown) indicated that only ABTS did not inactivate the enzyme. The reasons for inactivation are not clear given that other APxs do not show this behavior. However, it is proposed that it may be due to some sort of covalent modification of the enzyme due to a reaction with the oxidized aromatic radical.

2.4. Structural modeling

The structural model developed from rGmAPx (soybean) using the ascorbate bound structure (PDB ID: 1OAF) gives a good picture of the location of a key active site amino acid, R172 (Fig. 6). From this model, the S172 side-chain of the mutant would be right on the heme corner between the $\gamma$ and $\delta$ sides. If the aromatic binding site is indeed on the $\delta$ side of the heme, this would allow the serine side chain to have possible interaction with it. However, it clearly would be unable to interact with the ascorbate binding site in the same way as the WT arginine. As a result, the predicted activity with ascorbate would be significantly impacted. The observation of this impact on the actual expressed mutant indicates that our homology model for rPviAPx is useful to make structure/function predictions. Also, unlike a previously tested R172A mutant (Macdonald et al., 2006), the serine residue is still polar and able to hydrogen bond with a heme propionate group. The R172A mutant had a slightly reduced activity toward the aromatic substrate guaiacol whereas our R172S had a small increase in activity toward ABTS. Admittedly, this is not strong evidence of an influence on aromatic activity but the relatively small impact on aromatic activity fits with what is predicted from the homology model.

3. Conclusions

Overall, our data would suggest that switchgrass cytosolic PviAPx1 has structural and functional characteristics similar to those that have been studied crystallographically (rGmAPx and rPsAPx). The activity toward ascorbate and ABTS are also similar. rPviAPx does differ in that it appears to function as a monomer versus the homodimeric association of these other APxs previously reported. Yet it still displays non-Michaelis–Menten kinetics toward ascorbate. The R172S mutant indicated that ascorbate binding could be abolished without impairing peroxidase activity against other aromatic substances. This is not surprising given the strong structural evidence previously discussed for a separate aromatic binding site (see Section 1). The change made to generate the mutant was targeted to the proposed aromatic binding site and is far enough away from the aromatic binding site that it would not likely have any influence on activity toward aromatic substrates. It will be interesting to determine if the mutant protein can still play a role in antioxidant defense in plant cells under conditions of ascorbate limitations.
4. Experimental

4.1. Cloning

The open coding frame for a major cytosolic ascorbate peroxidase (rPviAPx; www.phytozome.org, Pavirv00022559m.g) was obtained from a cDNA library prepared from switchgrass callus tissue. The gene was fully sequenced and then cloned out of the library vector, pSPORT1 (Invitrogen). Degenerate PCR primers were used which added restriction enzyme sites to facilitate cloning and expression studies (NdeI (CATATG) to 5'-end and a EcoRI (GAATTC) at the 3'-end. Following sequence verification, the construct was cloned into the pET28a + plasmid (Invitrogen), behind a thrombin cleavable 6X-His tag to the N-terminal end. DNA sequencing was used to verify insertion frame and sequence integrity.

4.2. Homology modeling

Since there is no experimentally derived structural model currently available for rPviAPx, a homology model for the wild type enzyme was developed from the structure of recombinant soybean APx with ascorbate bound (PDB Id: 1OAF, Sharp et al., 2003) using UCSF Chimera (Pettersen et al., 2004) and Modeller (Sali and Blundell, 1993). This model was used to visualize the location of a key active site residue (R172) that interacts with ascorbate. An additional model was generated of a proposed single site APx mutant (R172S) to give a visual perspective of the size difference of the mutation and its location relative to the heme and ascorbate. All molecular images were generated from within Chimera using POV-Ray (Persistence of Vision Pty. Ltd. (2004) Persistence of Vision Raytracer (Version 3.6) retrieved from http://www.povray.org/download/).

4.3. Site-directed mutagenesis

Using pET28a.rPviAPx, site-directed mutagenesis was performed using mutant primers designed via the GeneTailor kit method (Invitrogen) that substituted the Arg172 codon (CGA) with a Ser codon (TCA). Platinum™ Taq DNA Polymerase High Fidelity (Invitrogen) was used with a protocol that had a 1 min, 56°C annealing step and a 9 min, 70°C elongation step. The R172S mutant was verified via DNA sequencing.

4.4. Expression and purification

For protein expression pET28a.rPviAPx was transformed into BL21(DE3) cells. Transformants were fully re-sequenced for rPviAPx and stored at −80°C as glycerol freezer stocks. These freezer stock were used to start overnight cultures with kanamycin (30 mg/L) which were then used to start expression cultures of 500 mL of 2x YT media in 1 L Erlenmeyer flasks containing kanamycin and 0.5 mM 5-aminolevulinic acid (Research Products International). The expression cultures were incubated at 31°C with shaking at 150 rpm for 24 h. These cultures were then induced at 0.5 mM IPTG and incubated at 37°C with shaking at 250 rpm for
6 h. Cells were harvested via centrifugation at which point the cell pellet was observed to have a dark brownish-red color. Pellet was resuspended in lysis buffer (50 mM NaPO₄, 500 mM NaCl, 10 mM imidazole, pH 8) to a concentration of 5 μg/mL and pellet was transferred to metal beaker. Fastbreak solution (Pierce) was added to beaker and solution was shaken at 4 °C for approximately 10 min. This generated very viscous cell lysate which was sonicated with an 11 mm tip using a Sonic Dismembrator – Model 500 (Fisher) until solution was non-viscous. The lysate was then centrifuged at 34,000g for 30 min which generated a very red cleared lysate and a fairly small white pellet with a small amount of red. The cleared lysate was filtered prior to purification using 5 mL FastFlow NTA columns (GE Healthcare), which had been freshly regenerated prior to use. Bound protein was washed first with lysis buffer until the A₂₈₀ came back to baseline. It was then washed with ten column volumes of wash buffer (50 mM NaPO₄, 500 mM NaCl, 25 mM imidazole, pH 8). Finally, the protein was eluted with elution buffer (50 mM NaPO₄, 500 mM NaCl, 250 mM imidazole, pH 8). Eluted protein was desalted using a series of 5 × 5 mL HiTrap desalting columns (GE Healthcare) into PBS (1 x). The desalted protein was treated with thrombin at 1 U/mg protein where protein concentration was calculated from E₂₈₀ (1.11 mL mg⁻¹ cm⁻¹). Background oxidation was measured by determining of rPviAPx.

Compound I was generated by addition of a molar equivalent of guaiacol (1 cm²/C₀) on a HPLC system. A 50 mM Tris–HCl, 100 mM KCl (pH 7) buffer was used for isocratic runs with a flow rate of 0.35 mL/min. Gel filtration molecular weight markers were normalized based on total numbers of transcripts within each dataset to allow comparisons between each experiment. These values are an approximate representation and not an absolute value.

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